Acta Crystallographica Section C Crystal Structure Communications

ISSN 0108-2701

Tb₂Ni₂Mg₃: a new structure type derived from the Ru₃Al₂B₂ type

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Received 15 December 2006 Accepted 11 January 2007 Online 31 January 2007

Single crystals of diterbium dinickel trimagnesium, Tb₂Ni₂Mg₃, were synthesized from the elements by induction melting. The novel compound crystallizes in the space group Cmmm with one Mg atom of site symmetry mmm and the Tb, Ni and other Mg atom in m2m positions. This ternary compound represents a new structure type that is derived from Ru₃Al₂B₂ by way of Wyckoff site distribution. The twolayer structure of Tb₂Ni₂Mg₃ is a new representative of a homologous linear structure series of general formula $R'_{k+n}X_{2n}R''_{2m+k}$ based on structural fragments of the α -Fe, CsCl and AlB₂ structure types. The Tb atoms in the structure are enclosed in 17-vertex polyhedra, while rhombododecahedra and distorted rhombododecahedra surround the Mg atoms, and equatorially tricapped trigonal prisms form around the Ni atoms. All interatomic distances indicate metallic type bonding.

Comment

Recently, rare-earth transition metal intermetallics have been studied because of their particular mechanical and physical properties as superconductors, batteries, magnets, magnetocaloric materials, *etc.* They also have great potential as hydrogen storage materials. Mg-modified alloys look to be particularly effective because of higher hydrogen capacity with lower density and cost. A number of recent studies of such materials have been reported. Kadir *et al.* (1997) have studied a series of hydrogen storage alloys of stoichiometry (RE)Ni₉Mg₂ (RE = La–Nd, Sm and Gd) that crystallize in the PuNi₃ structure. The hydrogen capacity of (RE)Ni₄Mg (RE = Y, La and Nd) compounds with the MgCu₄Sn structure type vary from 3.4 to 4.02 H atoms per formula unit under different conditions (Aono *et al.*, 2000; Guénée *et al.*, 2003). It was shown that the LaNi₂Mg ternary compound with the Heuslertype structure adsorbs hydrogen, forming the quaternary metal hydride LaMg₂NiH₇ (Renaudin *et al.*, 2003). Li *et al.* (2004) reported that the La_{1.5}Mg₁₇Ni_{0.5} alloy adsorbs 5.4 wt. % H at 573 K at excellent speed without any activation process. The electrochemical performances of a number of La–Ni–Mg alloys with high Ni content show promising electrode properties with higher discharge capacity (~400–410 mAh g⁻¹) than those employing alloys commonly adopted at present (Kohno *et al.*, 2000; Liao *et al.*, 2003).

The crystal structure, chemical bonding and physical properties of the series of RE2{Cu,Ni,Pd}2Mg ternaries with Mo₂FeB₂ structure type were investigated by Lukachuk & Pöttgen (2003). The new non-metallic La₂MgNi₂H₈ (1.89 wt. % H) complex hydride was obtained readily by hydrogenation of La₂MgNi₂ (Chotard et al., 2006). Detailed investigation of the constitutional properties of La-Ni-Mg at 673 K (De Negri et al., 2005) and La-Cu-Mg (De Negri et al., 2007) have indicated a great number of novel ternary phases that potentially could be hydrogen storage materials. Under such conditions, the accurate determination of the crystal structure of new intermetallics is the basic requirement for a better understanding of their physical properties. In this paper, the results of our recent structure investigation of the Tb₂Ni₂Mg₃ ternary compound, obtained during a systematic investigation of the Tb-Ni-Mg system at 673 K, are reported.

The Ru₃Al₂B₂ structure type was discovered by Jung & Schweitzer (1986). Since then, other representatives of this structure type among intermetallics have not been found. The title compound represents a new structure type that results from a site redistribution of the Ru₃Al₂B₂ structure, where the A1 (4*j*) site is now occupied by Mg atoms. Two Ru-atom positions (2*a* and 4*i* sites) are redistributed among Mg and Tb atoms, respectively. The B (4*j*) site corresponds to the Ni-atom positions. Thus, the compound could be represented also as (Tb₂Mg)Mg₂Ni₂. The unit-cell projection of Tb₂Ni₂Mg₃ on to the (001) plane with delineated coordination polyhedra is

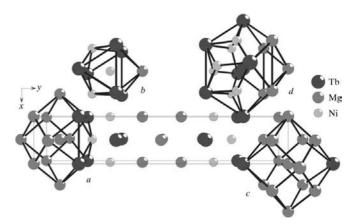


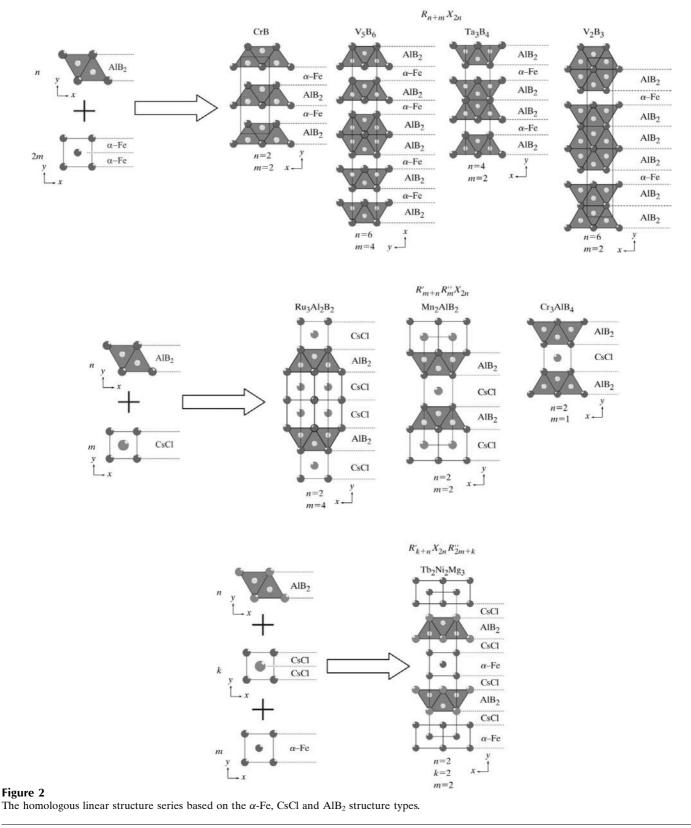
Figure 1

A perspective view of the unit-cell projection of $Tb_2Ni_2Mg_3$ on to the (001) plane. Coordination polyhedra of Tb (*d*), Mg1–Mg2 (*a* and *c*) and Ni (*b*) are shown.

inorganic compounds

shown in Fig. 1. The first coordination spheres [including bonding interactions up to 4 Å according to the empirical rule of $2^{1/2} \times \delta_{\min}$ of Krypyakevich (1977)] of the atoms are normal for intermetallics. The Tb atoms (site symmetry m2m) are surrounded in the structure by 17 adjacent atoms, *i.e.*

[(Tb1)Ni₆Mg₅Tb₆] (*d*). A rhombododecahedron [(Mg2)-Tb₂Mg₁₂] (*c*) and distorted rhombododecahedron [(Mg1)-Tb₄Mg₉Ni] (*a*) surround atoms Mg2 (*mmm*) and Mg1 (*m2m*), respectively. The Ni atoms (*m2m*, bonding interaction < 3.6 Å) are characterized by equatorially tricapped trigonal prisms



[(Ni1)Tb₆Ni₂Mg] (b). The structure of Tb₂Ni₂Mg₃ belongs to class No. 10 [coordination number 6 + n (n = 0-5) for the smallest atom, trigonal prism and its derivatives as coordination polyhedra] according to the classification scheme of Krypyakevich (1977). The interatomic distances (Table 1) are in good correlation with the sums of the atomic radii (Emsley, 1991) and indicate metallic type bonding (the shortest Tb—Ni distance is 2.812 Å, which is 93.4% of the sum of the atomic radii).

According to Krypyakevich (1977), structure types belong to homologous series if they are composed of the same multiple fragments and differ between themselves by quantitative ratios of these fragments, so that a general formula and ratio describes all structures of the homologous series. The linear structure series is assigned when the two-dimensional segments (infinite slabs) of constituent fragments are stacked one-dimensionally along a stacking direction. Kuz'ma (1983) has described a homologous linear structure series of borides of general formula $R'_{n+m}X_{2n}$, with constituent fragments of the α -Fe and AlB₂ structure types (*n* is the number of blocks of AlB₂-type trigonal prisms and *m* is the number of half unit-cell blocks of α -Fe structure type). This series (Fig. 2) includes the CrB (n = 2 and m = 2), V₅B₆ (n = 6 and m = 4), Ta₃B₄ (n = 4)and m = 2) and V₂B₃ (n = 6 and m = 2) structure types. Another homologous linear structure series consists of fragments of the CsCl and AlB₂ structure types (Fig. 2) with formula $R'_{n+m}R''_mX_{2n}$ (*n* is the number of blocks of AlB₂-type trigonal prisms and m is the number of blocks of CsCl-type cubes), including $Ru_3Al_2B_2$ (n = 2 and m = 4), Mn_2AlB_2 (n = 2and m = 2) and Cr₃AlB₄ (n = 2 and m = 1). Tb₂Ni₂Mg₃ can be viewed as the first representative of a novel linear structure series based on the α -Fe, CsCl and AlB₂ structure types combined. The general formula of this series is $R'_{k+n}X_{2n}R''_{2m+k}$ (*n* is the number of blocks of AlB_2 -type trigonal prisms, *m* is the number of unit-cell blocks of α -Fe structure type and k is the number of blocks of CsCl-type cubes). For Tb₂Ni₂Mg₃ itself, m = n = k = 2. Tb₂Ni₂Mg₃ belongs to the family of twolayer compounds, along with more than 70 other inorganic structure types (mainly borides) according to TYPIX (Parthé et al., 1993-1994).

Experimental

The alloys were prepared by induction melting of stoichiometric amounts of the constituent metals (purity > 99.9% for Tb and Mg, and 99.98% for Ni) placed together into outgassed tantalum crucibles sealed by arc-welding under conditions of pure argon. The alloys were then annealed in an evacuated quartz tube at 673 K for three weeks. After annealing, the samples could be readily separated from the tantalum crucible. No side reaction of the samples with the crucible was detected. The samples, in the form of compact buttons as well as fine-grained powders, were stable against air and moisture influence. The sample preparation method is completely reproducible. A series of alloys containing the Tb₂Ni₂Mg₃ phase were analyzed by scanning electron microscopy and energy-dispersive X-ray spectroscopy (EDXS) in order to study their microstructure and phase composition (with ± 0.5 at.% accuracy). A single crystal was extracted from an alloy of nominal composition Tb₂₇Ni₂₂Mg₅₁ (consisting mainly of the $Tb_2Ni_2Mg_3$ phase). An average result of the EDXS analysis for the bulk samples is 29.6 at.% Tb, 29.0 at.% Ni and 41.4 at.% Mg, which compares well with the composition obtained from the structural refinement.

Crystal data

Tb1-Mg2

Crystal data			
Tb ₂ Ni ₂ Mg ₃ $M_r = 508.19$ Orthorhombic, <i>Cmmm</i> a = 3.9788 (7) Å b = 21.203 (4) Å c = 3.6583 (7) Å V = 308.62 (10) Å ³ Z = 2		$D_x = 5.469 \text{ Mg m}^{-3}$ Mo K\alpha radiation $\mu = 28.80 \text{ mm}^{-1}$ T = 295 (2) K Irregularly shaped, metallic dark grey 0.1 \times 0.08 \times 0.04 mm	
Data collection			
Oxford Diffraction Xcalibur diffractometer ω scans Absorption correction: analytical (<i>CrysAlis RED</i> ; Oxford Diffraction, 2005) $T_{min} = 0.075, T_{max} = 0.313$		940 measured reflections 213 independent reflections 199 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.043$ $\theta_{\text{max}} = 26.4^{\circ}$	
Refinement			
Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.032$ $wR(F^2) = 0.078$ S = 1.10 213 reflections 16 parameters		$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0609P)^{2}]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 2.23 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -1.21 \text{ e } \text{\AA}^{-3}$	
Table 1 Selected bond lengths	s (Å).		
Tb1-Ni1 ⁱ	2.8118 (7)	Tb1-Tb1 ^{iv}	3.9788 (7)
Tb1-Ni1	2.9735 (17)	Ni1-Ni1 ⁱ	2.533 (3)
Tb1-Mg1 ⁱ	3.386 (4)	Ni1-Mg1	2.817 (6)
Tb1-Tb1 ⁱⁱ	3.6583 (7)	$Mg1 - Mg2^{v}$	3.193 (3)
Tb1-Tb1 ⁱⁱⁱ	3.7007 (13)	Mg1-Mg1 ^{vi}	3.400 (12)

Symmetry codes: (i) $-x + \frac{1}{2}, -y + \frac{1}{2}, -z + 1$; (ii) x, y, z + 1; (iii) $-x - \frac{1}{2}, -y + \frac{1}{2}, -z$; (iv) x - 1, y, z; (v) $x + \frac{1}{2}, y + \frac{1}{2}, z + 1$; (vi) -x, -y + 1, -z + 1.

3.7405 (9)

Systematic absences indicated the possible space groups C222 (No. 21), Cmm2 (No. 35), Amm2 (No. 38) and centrosymmetric Cmmm (No. 65). A statistical test of the distribution of the E values (Farrugia, 1999) suggested that the structure is non-centrosymmetric; however, following the advice of Marsh (1995), the structure solution and refinement were also performed in the centrosymmetric group. The results clearly indicated that Tb₂Ni₂Mg₃ crystallizes in the centrosymmetric space group Cmmm, since the solution and refinement in the non-centrosymmetric variants were not satisfactory. The occupancy parameters were refined for this structure in order to check for deviations from the ideal composition. No significant deviations were found. Hence, in the final refinement cycles of this structure, the ideal occupancy parameters were retained. Data were then refined with anisotropic displacement parameters for all atoms. The atomic coordinates were standardized using the STRUCTURE TIDY program (Gelato & Parthé, 1987). The final difference Fourier syntheses revealed no significant residual peaks; the highest maximum residual electron density is 2.23 Å from Tb and the deepest hole is 1.21 Å from Ni.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2004); cell refinement: *CrysAlis CCD*; data reduction: *CrysAlis RED* (Oxford Diffraction, 2005); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97*

(Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *SHELXL97*.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: SQ3057). Services for accessing these data are described at the back of the journal.

References

- Aono, K., Orimo, S. & Fujii, H. (2000). J. Alloys Compd, 309, L1-L4.
- Brandenburg, K. (1999). *DIAMOND*. Version 2.1e. Crystal Impact GbR, Bonn, Germany.
- Chotard, J.-N., Filinchuk, Ya., Revaz, B. & Klaus, Y. (2006). Angew. Chem. Int. Ed. 45, 7770–7773.
- De Negri, S., Giovannini, M. & Saccone, A. (2005). J. Alloys Compd, 397, 126– 134.
- De Negri, S., Giovannini, M. & Saccone, A. (2007). J. Alloys Compd, **427**, 134–141.
- Emsley, J. (1991). The Elements, 2nd ed. Oxford: Clarendon Press.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Gelato, L. M. & Parthé, E. (1987). J. Appl. Cryst. 20, 139-143.

- Guénée, L., Favre-Nicolin, V. & Yvon, K. (2003). J. Alloys Compd, 348, 129– 137.
- Jung, W. & Schweitzer, K. (1986). Z. Kristallogr. Kristallgeom. Kristallphys. Kristallchem. 174, 109–110.
- Kadir, K., Sakai, T. & Uehara, I. (1997). J. Alloys Compd, 257, 115-121.
- Kohno, T., Yoshida, H., Kawashima, F., Inaba, T., Sakai, I., Yamamoto, M. & Kanda, M. (2000). J. Alloys Compd, **311**, L5–L7.
- Krypyakevich, P. I. (1977). Structure Types of Intermetallic Compounds. Moscow: Nauka. (In Russian.)
- Kuz'ma, Yu. B. (1983). Crystallochemistry of Borides. Lviv: Vyshcha Shkola. (In Russian.)
- Li, Q., Lin, Q. & Chou, K.-C. (2004). J. Alloys Compd, 368, 101-105.
- Liao, B., Lei, Y., Lu, G., Chen, L., Pan, H. & Wang, Q. (2003). J. Alloys Compd, 356–357, 746–749.
- Lukachuk, M. & Pöttgen, R. (2003). Z. Kristallogr. 218, 767-787.
- Marsh, R. E. (1995). Acta Cryst. B51, 897-907.
- Oxford Diffraction (2004). *CrysAlis CCD*. Version 1.170. Oxford Diffraction Ltd, Abingdon, Oxfordshire, England.
- Oxford Diffraction (2005). CrysAlis RED. Version 1.171. Oxford Diffraction Ltd, Abingdon, Oxfordshire, England.
- Parthé, E., Gelato, L., Chabot, B., Penzo, M., Cenzual, K. & Gladyshevskii, R. (1993–1994). TYPIX – Standardized Data and Crystal Chemical Characterization of Inorganic Structure Types, in Gmelin Handbook of Inorganic and Organometallic Chemistry. Heidelberg: Springer.
- Renaudin, G., Guénée, L. & Yvon, K. (2003). J. Alloys Compd, 350, 145–150. Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.